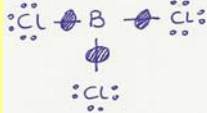
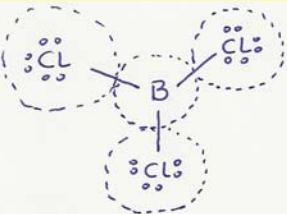


**Molecular Shape: EXERCISES**

Form.	Lewis and electron groups	A-X-E	Electron groups	Atoms SHAPE	Molecule
<b>BCl<sub>3</sub></b>		<b>AX<sub>3</sub></b>	<b>3 groups 120°</b>	<b>4 atoms TRIGONAL PLANAR</b>	
<b>CH<sub>4</sub></b>					
<b>CHCl<sub>3</sub></b>					
<b>H<sub>2</sub>CO</b>					
<b>HCN</b>					
<b>CO<sub>2</sub></b>					
<b>CCl<sub>4</sub></b>					

$\text{NH}_3$					
$\text{NH}_4^+$					
$\text{H}_2\text{O}$					
$\text{SO}_2$					
$\text{SO}_3$					
$\text{H}_2\text{S}$					