
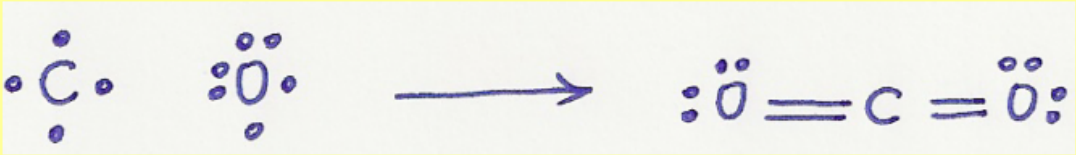
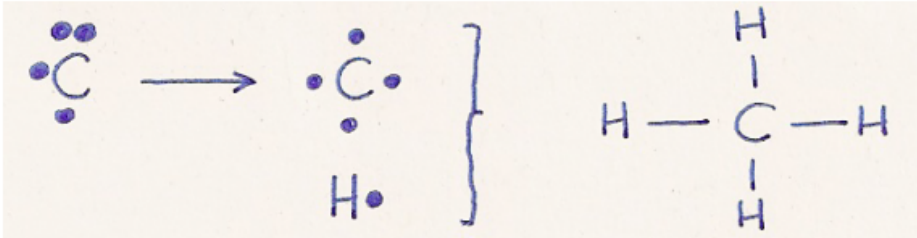
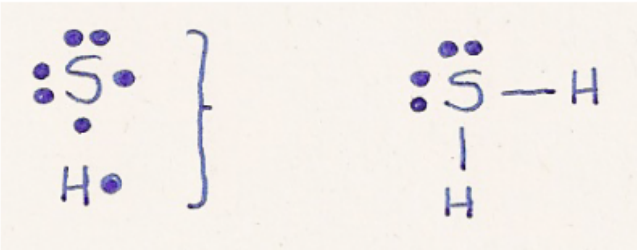


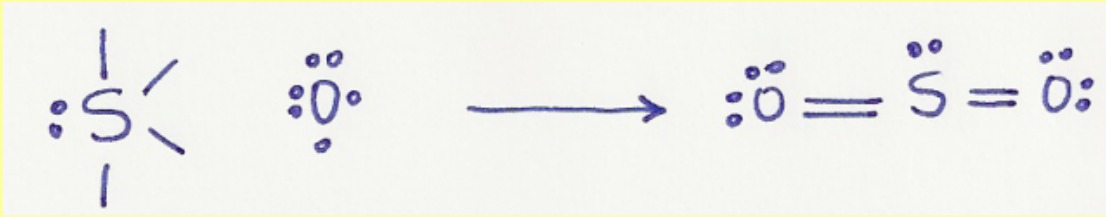
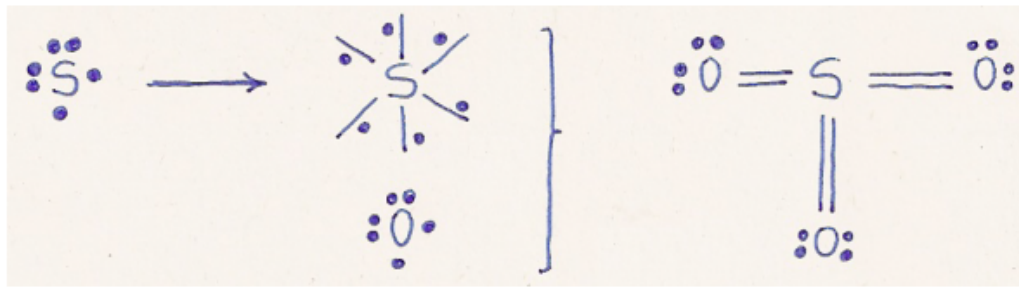
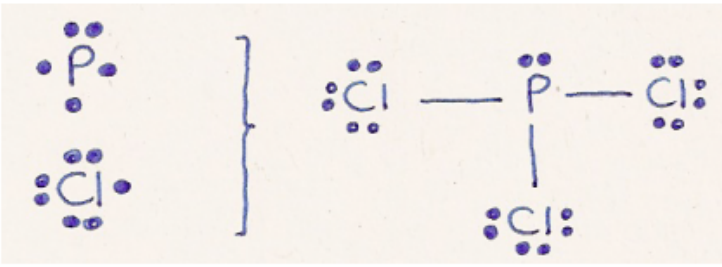
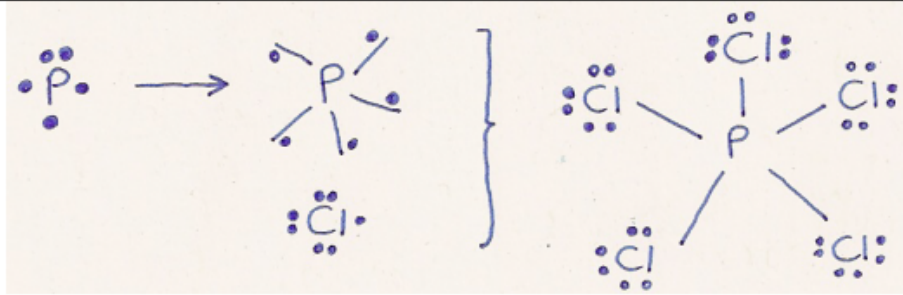
Covalent Bonding: solutions

| EXERCISE: Draw the following molecules, using the Lewis notation | |
|---|--|
| NH₃ |  |
| CO₂ |  |
| CH₄ |  |
| H₂S |  |

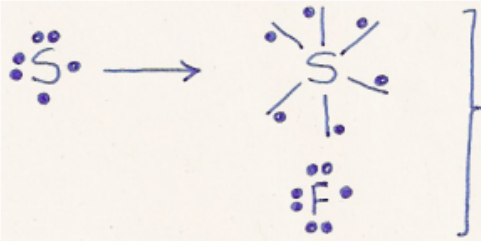
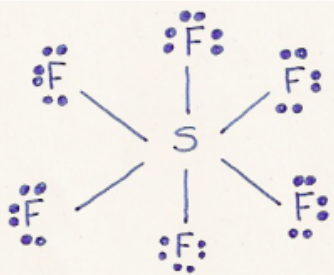
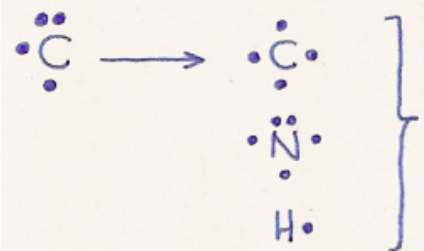
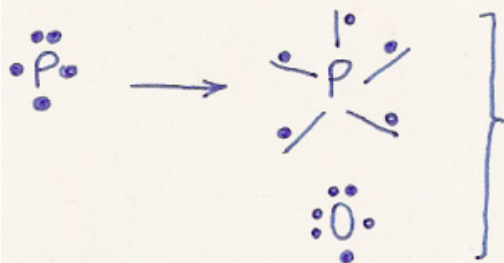
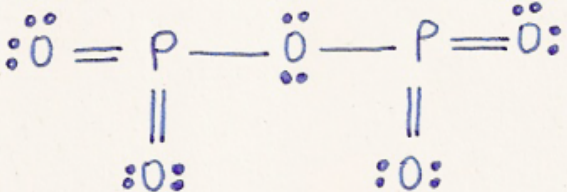
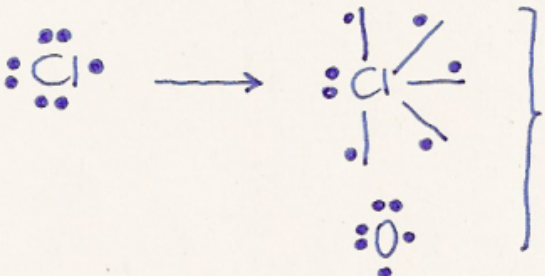
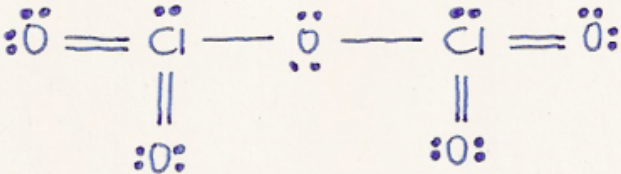
Covalent Bonding: solutions

| | |
|-----------------------|--|
| CO | |
| BF₃ | |
| HCl | |

Covalent Bonding: solutions

| | |
|------------------------|--|
| SO₂ |  |
| SO₃ |  |
| PCl₃ |  |
| PCl₅ |  |

Covalent Bonding: solutions

| | | |
|---|--|---|
| <p>SF₆</p> |  |  |
| <p>HCN</p> |  | <p>H—C≡N</p> |
| <p>P₂O₅</p> |  |  |
| <p>Cl₂O₅</p> |  |  |