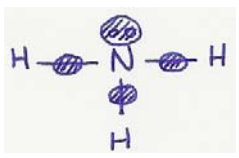

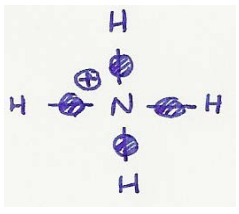
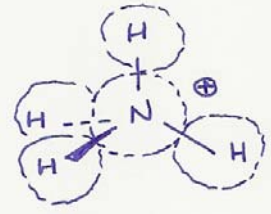
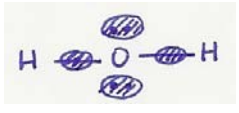

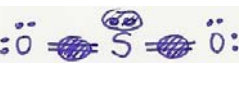

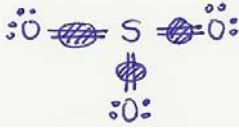
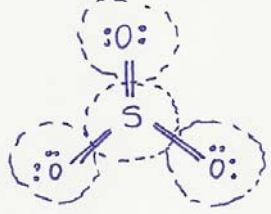



Molekularen geometriak: EMAITZAK

Form.	Lewis eta talde elektronikoak	A-X-E	Elektroi taldeak	Atomoak GEOMETRIA	Molekula
BCl_3		AX_3	3 talde 120°	4 atomo TRIGONAL LAUA	
CH_4		AX_4	4 talde 109°	5 atomo TETRAEDRIKOA	
CHCl_3		AX_4	4 talde 109°	5 atomo TETRAEDRIKOA	
H_2CO		AX_3	3 talde 120°	4 atomo TRIGONAL LAUA	
HCN		AX_2	2 talde 180°	3 atomo LINEALA	
CO_2		AX_2	2 talde 180°	3 atomo LINEALA	
CCl_4		AX_4	4 talde 109°	5 atomo TETRAEDRIKOA	

NH_3		AX_3E	4 talde 109°	4 atomo PIRAMIDALA	
NH_4^+		AX_4	4 talde 109°	5 atomo TETRAEDRIKOA	
H_2O		AX_2E_2	4 talde 109°	3 atomo ANGELUARRA	
SO_2		AX_2E	3 talde 120°	3 atomo ANGELUARRA	
SO_3		AX_3	3 talde 120°	4 atomo TRIGONAL LAUA	
H_2S		AX_2E_2	4 talde 109°	3 atomo ANGELUARRA	